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CZECHOSLOVAKIA

DOLIHAL, J; TOVAREK, J; POKORNY, J.

Third Internal Medicine Clinic of the Medical Faculty of  
J. E. Purkyne University (III. vnitřní klinika lékařské  
fakulty University J. E. Purkyne), Brno (for all)

Prague, Vnitřní lékařství, No 10, 1963, pp 1010-1013

"The Clinical Picture of Acute Intermittent Porphyria."

POJER, J., prof. MUDr.; MASTIK, C.; NINGER, E.; DOLIHAL, J. Technicka  
spoluprace: TOVAREK, J.; POKORNY, J.

Hepatic changes in heart infarction. Bratisl. lek. listy 45  
no. 8: 460-468 30 Ap '65.

1. III. vnitrni klinika Lekarske fakulty University J.E. Purkyne  
v Brne (veduc.: prof. MUDr. J. Pojer).

DOLIN, A.O.		PROCESSING AND PRIORITIZATION INDEX	
<p>The role of the cerebral cortex in the regulation of metabolic processes. A. O. Dolin, E. T. Minker-Bogdanova and Yu. A. Povornikoff. Arch. sci. Med. (U. S. S. R.) 268, 66-77 (in English 77) (1934).—The sugar, Cl and fat curves were detd. in the blood of 8 female patients with hysteria under hypnosis suggestion of receiving sugar when water was actually given and vice versa. It is claimed that the suggested idea rather than the substance given detd. the shape of the blood-sugar curves, but the cited figures are contradictory and are not convincing.</p> <p>W. A. Perlswig</p>			
ASS. SLA METALLURGICAL LITERATURE CLASSIFICATION			
SEARCHED		SERIALIZED	
INDEXED		FILED	
MAY 1964		MAY 1964	

DOLIN A. O.

DOLIN A. O.

Generalizatsiia uslovnykh reflektsov v period ikh spetsializatsii kak forma protivleniia sistemnosti korikovykh protsessov. /Close correlation of elements of a complex system and correlation of the entire complex with surrounding conditions/ Tr. Fisiol. laborat. Pavlova 16: 1949 p. 360-81.

1. Of the Physiological Division of the All-Union Institute of Experimental Medicine (Head — Academician I. P. Pavlov). CML Vol. 19, No. 1 July 1950

DOLIN, A.O.

[Summation of internal inhibition and its role in the modification of unconditioned alimentary, and defense reflexes] Summatsiia vnutrennego tormosheniia i ee rol' v izmenenii bezuslovnykh pishchevykh i oboronitel'nykh refleksakh. Tr.Fiziol.laborat.Pavlova 16:382-397 '49. (CML 19:1)

1. Of the Physiological Department imeni Academician I.P.Pavlov of the All-Union Institute of Experimental Medicine and the Department of the Physiology and Pathology of Higher Nervous Activity imeni Academician I.P.Pavlov of the State Institute for the Advanced Training of Physicians.

DOLIN, A.O.

Conditioned reflex cataleptic condition. Zh. vysshei nerv. deiat.  
Pavlova 1 no.4:485-499 July-Aug 1951. (CIME 23:2)

1. Laboratory of Experimental Neuropathology of the Physiology Department  
imeni Academician I. P. Pavlov of the All-Union Institute of Experimental  
Medicine and Department of the Physiology and Pathology of Higher Nervous  
Activity imeni Academician I. P. Pavlov, State Institute for the Ad-  
vanced Training of Physicians.

DOLIN, A.O.

Pathological and protective role of conditioned and unconditioned reactions. Zh. vysshei nerv. deiat. 1 no. 6:934-943 Nov-Dec. 1951.

(GLML 23:3)

1. Moscow.

DOLIN, A. G.

DOLIN, A. G. -- "Conditioned-Reflex Emergence and Suppression of  
Pathological Conditions in an Organism." Sub 22 Apr 52, Central Inst  
for the Advanced Training of Physicians. (Dissertation for the Degree  
of Doctorate in Medical Sciences) .

SC: Vechernaya Moskva January-December 1952



DOLIN, A.O.; ZBOROVSKAYA, I. I.

Further study on the process of inhibition (induction inhibition)  
in repression of pathologic conditions. Zh. vysshei nerv. deiat.  
2 no. 1:70-77 Jan-Feb 1952. (CIAM 23:3)

USSR/Medicine - Immunology

Jul/Aug 52

"Role of the Cortex in Immune Reactions of the Organism," A. O. Dolin, V. N. Krylov, Mil Med Acad J, 1952, S. M. Kirov and Chair of Physiol and Pathol of Higher Nervous Activity, Cen Inst for Advanced Trng of Physicians

"Zhur Vyssh Nerv Deyat" Vol 2, No 4, pp 547-560

Results of expts, particularly when dynamic stereotypes are established, offer sufficient ground to think that in immune reactions, as in defensive reactions of an organism, cortical dependence is extremely

234T44

intensive and firm. Exptl data obtained by inducing immunological and toxic reactions with the aid of conditional irritants furnishes the basis for confirmation that the cortical component is necessarily a part of the over-all phenomena which form the immunological reaction, strengthening or weakening it depending on intercentral relationships that are formed. All this attests to the significance and specificity of expts which were directed toward the study of the defensive, protective role of the nervous system and the particularly great part that the cortex plays in that activity.

234T44

DOLIN, A. O.

DOMIN, A.O.

50 years of the theory on conditioned reflexes. Sovet. sdravookur.  
11 no.2:3-9 Mar-Apr 1952. (GML 22:1)

1. Professor. 2. Moscow.

LEPESHINSKAYA, O.B., professor; USIYEVICH, M.A., professor; ASRATYAN, E.A., professor; SMIRNOV, A.I., professor; FILIPPOVICH, S.I., doktor meditsinskikh nauk; VOLOKHOV, A.A., professor; FILIMONOV, I.N., professor; SHTAKIN, P.G., professor; CHERNIGOVSKIY, V.N., professor; SPERANSKIY, A.D., akademik; DOLIN, A.O., doktor meditsinskikh nauk; KOTLYAREVSKIY, L.I., professor; ~~MEGOVSKIY~~ V.A., professor; KASATKIN, N.I., professor; STEL'CHUK, I.V., professor; YEGOROV, B.G., professor; BAKULEV, A.N., professor; SMIRNOV, L.I., professor; USPENSKIY, V.N., redaktor; PETROV, S.P., redaktor.

[Teachings of I.P.Pavlov in theoretical and practical medicine]

Uchenie I.P.Pavlova v teoreticheskoi i prakticheskoi meditsine. Vol.2. Moskva, Izd-vo Ministerstvo zdoravookhraneniya SSSR, 1953. 611 p.

(MLRA 7:3)

1. Deystvitel'nyy chlen AMN SSSR (for Lepeshinskaya, Chernigovskiy and Bakulev).
2. Chlen-korrespondent Akademii nauk SSSR (for Asratyan).
3. Chlen-korrespondent AMN SSSR (for Smirnov, Filimonov, Yegorov and L.I.Smirnov).
4. Moscow. Tsentral'nyy institut usovershenstvovaniya vrachey. (Pavlov, Ivan Petrovich, 1849-1936) (Nervous system) (Physiology)

DOLIN, A.O.

Chem Obs V48

1-25-54

Pathology

Characterization test of epileptiform state phases by means of the transhemation reaction. I. A. Blyumenfeld, A. O. Dolin, S. E. Krasovskaya, T. P. Kolich, and A. M. Charnyi (Ministry Health, Moscow). *Doklady Akad. Nauk S.S.S.R.* 92, 189-91(1953).—The rate of the transhemation reaction reflects the phases of the epileptic state. Preconvulsion state reduces the rate of transhemation, a change that becomes evident immediately upon introduction of camphor. The retardation is then replaced with a constantly increasing rate of transhemation. If the convulsions are feeble, the latter phase is not apparent. The postconvulsive state is characterized by gradual retardation of transhemation. The expts. were run on rabbits, and the transhemation reaction was run as described earlier (C.A. 46, 76377).

G. M. Kosolapoff

(6)

DOLIN, A.O.; SALGANNIK, G.M. [deceased]

Physiological principles of psychoprophylactic preparation for  
labor. Akush. i gin. 32 no.3:47-62 Ny-Je '56. (MLRA 9:9)

1. Iz Instituta kusherstva i ginekologii (dir. L.G.Stepanov)  
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(LABOR,

painless, psychoprophylactic method, physiol. aspects  
(Russ)

DOLIN, A.O., (Moskva)

Experimental pathology of the higher nervous activity. Zhur.vys.  
nerv.deiat. 7 no.6:819-830 M-D '57. (MIRA 11:2)  
(CENTRAL NERVOUS SYSTEM, physiology,  
higher nervous activity, exper. disord., review (Rus))

ANOKHIN, P.K., BABSKIY, Ye.B., VERESHCHAGIN, N.K., KABANOV, A.N., PUGHEOV, N.V.  
DOLIN, A.O.

Nikolai Vladimirovich Timofeev; an obituary. Fiziol. zhur. 44  
no.9:904-905 S '58 (MIRA 11:12)  
(TIMOFEEV, NIKOLAI VLADIMIROVICH, 1896-1958)



DOLIN, A.O., (Moskva)

Nature of the double role of inhibition as compared with  
reactions of the organism to chemical epileptogenic stimuli.  
Zhur.vys.nerv.deiat. 9 no.3:343-353 My-Je '59. (MIRA 12:9)  
(REFLEX, CONDITIONED)  
(CONVULSIONS, experimental)

DOLIN, A.O.; KRYLOV, V.N.; LUK'YANENKO, V.I.; FLEROV, B.A.

Recent experimental data on conditioned reflex production and the inhibition of immune and allergic reactions. Zhur.vys.nerv.deiat. 10 no.6:832-841 N-D '60. (MIRA 14:1)

1. Kafedra fiziologii vysshey nervnoy deyatel'nosti Moskovskogo gosudarstvennogo universiteta im. M.V.Lomonosova.  
(CONDITIONED RESPONSE) (ALLERGY) (IMMUNITY)

~~DOLIN, Aleksandr Osipovich~~ prof.; KOZAROVITSKIY, L.B., red.;  
KAPISHEVA, V.S., red. izd-va; GRIGORCHUK, L.A., tekhn.  
red.

[Pathology of the higher nervous activity] Patologiya vys-  
shei nervnoi deiatel'nosti. Moskva, Gos.izd-vo "Vysshaya  
shkola," 1962. 338 p. (MIRA 16:8)  
(BRAIN—DISEASES)

DOLIN, A.O. (Moskva); FARBER, D.A. (Moskva); ZMANOVSKIY, Yu.F. (Moskva)

Determination of the lability of nervous processes based on the reaction of assimilating the rhythm of light flickering. Zhur. vys. nerv. čelst. 15 no.2:381-392 Mr-Apr '65.

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DOLIN, B.

Hunting with a movie camera. Moskva, Goskinoizdat, 1951.

SHAROV, I., inzh.; DOLIN, I., inzh.

Using plastic materials in building apartment houses. Zhil.-  
kom.khoz. 9 no.6:17-18 '59. (MIRA 12:10)  
(Moscow--Apartment houses) (Plastic materials)

DOLIN, I. I. and BRUSENITSEVA, S. A. (Inst of Physical Chemistry)

"The Action of Ionizing Radiation on Concentrated Water Solutions of Inorganic Substances"

Isotopes and Radiation in Chemistry, Collection of Papers of 2nd All-Union Sci-Tech. Conf. on Use of Radioactive and Stable Isotopes and Radiation in National Economy and Science, Moscow, 12d-vol. AN SSSR, 1958, 360pp.

This volume publishes the reports of the Chemistry Section of the 2nd All-Union Sci-Tech Conf on Use of Radioactive and Stable Isotopes and Radiation in Science and the National Economy, sponsored by Acad. Sci. USSR and Main Admin for Utilization of Atomic Energy under Council of Ministers USSR, Moscow, 4-12 April 1957.

DOLIN, K.

Study, constructive work, relaxation. Prof.-tekhn. obr. 15 no. 7:14  
Jl '58. (MIRA 11:7)  
(Railroads--Employees--Education and training.)



9,3700

S/141/61/004/005/015/021  
E039/E120

AUTHOR: Dolin, L.S.

TITLE: On the possibility of comparison of three-dimensional electromagnetic systems with non-uniform anisotropic filling

PERIODICAL: Izvestiya vysshikh uchebnykh zavedeniy,  
Radiofizika, v.4, no.5, 1961, 964-967

TEXT: Electromagnetic field systems can be investigated without a direct solution of Maxwell's equations, by comparison with other simpler field systems. In this paper the possibility of comparing analogous three-dimensional systems is examined. Starting from Maxwell's equations in generalised orthogonal coordinates, invariant equations are derived relating electric and magnetic current densities, charge density, permittivity and permeability. These equations can be applied to a three-dimensional field system with anisotropic filling. A simple example is used to illustrate the method; this is a naturally oscillating resonator with perfectly conducting walls. It is shown that, by substitution, it is possible to examine the

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On the possibility of comparison of .. S/141/61/004/005/015/021  
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electromagnetic field components in a second resonator with different parameters for the filling medium. It should be noted that as it is not permissible to transform the properties of a medium independently, because there are well defined relations between them, this puts some limitation on the method. However, it is shown that the method is useful for the solution of electromagnetic field problems. ✓  
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There are 2 figures and 3 Soviet-bloc references.

ASSOCIATION: Nauchno-issledovatel'skiy radiofizicheskiy institut  
pri Gor'kovskom universitate  
(Scientific Research Institute for Radiophysics at  
Gor'kiy University)

SUBMITTED: March 11, 1961

Card 2/2

DOLIN, L.S.

Scattering of a light beam in a layer of a turbid medium.

Izv. vys. ucheb. zav. radiofiz. 7 no.2:380-382 '64 (MIRA 18:1)

1. Nauchno-issledovatel'skiy radiofizicheskiy institut pri  
Gor'kovskom universitete.

DOLIN, L.S.

Description of the radiation from weakly inhomogeneous wave fields.  
Izv. vys. ucheb. zav.; radiofiz. 7 no.3:559-562 '64. (MIRA 17:11)

1. Nauchno-issledovatel'skiy radiofizicheskiy institut pri Gor'kovskom  
universitate.

1.6712-65 EWA(k)/EWT(l)/WG(k)/EPA(s)-2/EEC(k)-2/K/EPA(s)-2/EEC(t)/T/  
EEC(b)-2/EMP(k)/EWA(m)-2 Po-4/Pz-6/Pab-24/Pf-4/Pi-4/P1-4 IJP(c)/ASD(s)-5/  
ESD(gs)/ESD(t)/RAEM(t) W/JEB/AT

ACCESSION NR: AP4044101

S/0141/64/007/003/0471/0478

AUTHOR: Dolin, I. S.

TITLE: Reciprocity relations for partially coherent electromagnetic fields

SOURCE: IVUZ. Radiofizika, v. 7, no. 3, 1964, 471-478

TOPIC TAGS: correlation analysis, reciprocity relation, Maxwell equation, electromagnetic diffraction, radiation resistance

ABSTRACT: Reciprocity relations are obtained for the tensor correlation functions for a random stationary electromagnetic field, first introduced by B. Wolf (Nuovo Cimento v.12, 884, 1954). Since these correlation functions have been shown to satisfy a system of linear differential equations equivalent to the Maxwell equations, the author considers it of interest to generalize the reciprocity principle, formulated originally for the case of regular fields, and

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ACCESSION NR: AP4044101

to change over to a correlation formulation of the reciprocity principle. It is shown that the electrodynamic reciprocity principle can be formulated in the form of relations not between the fields and the currents but between the correlation functions directly. He then proceeds to apply the results to the diffraction of random fields, using a modification of the method of auxiliary sources, which is widely used in the theory of radiation and diffraction of regular fields. As a particular example, the author considers the diffraction of a statistically homogeneous and isotropic field by an arbitrary object, where the problem of determining the correlation characteristics reduces to the determination of the mutual radiation resistance of the body and a set of elementary dipoles. The case when the diffracted field can be regarded as homogeneous over the length of the diffracting object is also treated from the reciprocity point of view. "In conclusion, I thank M. A. Miller for valuable advice and remarks." Orig. art. has: 24 formulas.

Cord 2/3

L 6712-65

ACCESSION NR: AP4044101

ASSOCIATION: Nauchno issledovatel'skiy radiofizicheskiy institut  
pri Gor'kovskom universitete (Scientific Research Radiophysics  
Institute at the Gor'kiy University)

SUBMITTED: 19Jul63

ENCL: 00

SUB CODE: EM

NR REF SOV: 003

OTHER: 007

Card 3/3

Z 15993-66 EEC(k)-2/ENT(d)/ENT(1)/FCC GW/WH-2/LHB  
ACC NR: AP6007631 SOURCE CODE: UR/0141/66/009/001/0061/0071

AUTHOR: Dolin, L. S.

ORG: Scientific Research Radiophysics Institute of Gor'kiy University (Nauchno-<sup>46</sup>  
issledovatel'skiy radiofizicheskiy institut pri Gor'kovskom universitete) <sup>B</sup>

TITLE: Propagation of a narrow light beam in a medium with pronounced anisotropic scattering <sup>21,44,55</sup>  
<sub>46</sub>

SOURCE: IVUZ. Radiofizika, v. 9, no. 1, 1966, 61-71

TOPIC TAGS: light propagation, searchlight sounding, light scattering, atmospheric optics

ABSTRACT: The problem of the propagation of an initially narrow, sharply directed light beam in a turbid medium with pronounced anisotropic scattering is examined. Proceeding from the transport equation in the approximation of small angles, the dependence of the  $x$ -axis illumination intensity, the dispersion of the angle of beam arrival, and the effective beam cross section on the optical path, initial beam characteristics, and angular width of the indicatrix of scattering is computed. The computations yield a reasonably complete qualitative picture of the structure of the beam in the area relatively near the source. The Card 1/3

UDC: 535.36

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ACC NR: AP6007631

basic characteristics of the beam with zero initial dispersion is shown diagrammatically (see Fig. 1). The upper curve depicts the axial distribution of the

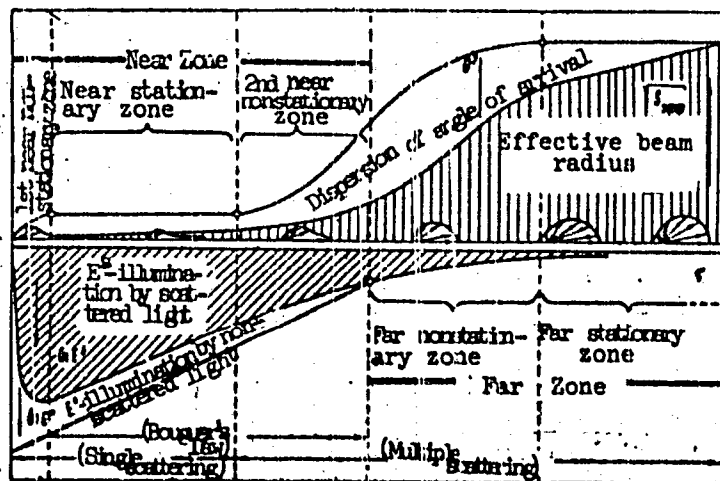


Fig. 1. Beam dispersion characteristics

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ACC NR: AP6007631

angular dispersion of beam arrival ( $\theta^2$ ). The upper lined area corresponds to the configuration of the beam, whose limits are set by a definite level of intensity fall-off. The deformation of the brightness body on the beam axis is illustrated by several qualitative diagrams. The lower half of the diagram shows the axial distribution of illuminations  $E^0$  and  $E^S$  by nonscattered and scattered light respectively. The beam is broken down into the zones indicated on the basis of the illumination intensities created on the beam axis by the nonscattered and scattered fields. The scale of the diagram is arbitrarily chosen. Orig. art. has: 21 formulas and 6 figures. [DM]

SUB CODE: 04/ SUBM DATE: 03May65/ ORIG REF: 009/ OTH REF: 002/ ATD PRESS: 4203

Card 3/3

AUTHORS: Dolin, N.F., and Zyabkina, Z.F. (Engineers) SOV/110-59-9-22/22  
TITLE: A Scientific-technical Conference on Rectifiers  
PERIODICAL: Vestnik elektropromyshlennosti, 1959, Nr 9, p 80 (USSR)  
ABSTRACT: A scientific-technical conference on rectifiers was held in Zaporozh'ye. There were more than 300 delegates from 135 organisations representing research institutes, and various organisations concerned with the development, manufacture, erection and operation of rectifiers. Expected developments in rectifier technique during the period 1959-65 were considered. The need for suitable rectifiers for many branches of industry was noted, but the present state of affairs was not considered satisfactory. New types of rectifiers including high-quality semiconductor materials are being introduced very slowly. In recent years research laboratories and factories have nevertheless achieved appreciable successes. A series of pumpless rectifiers is manufactured. Special types of rectifier equipment have been developed for different applications. Germanium rectifiers have been developed for currents from 10 to 200 A and voltages from 100 to 200 V. The ionic excitation system of the

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SOV/11C-59-9-22/22

A Scientific-technical Conference on Rectifiers

hydro-alternators at the Volga Power Station is working satisfactorily. Construction has commenced on the first full-scale 800 kV d.c. transmission line from the Stalingrad power station to the Donbas. Sixty reports and communications were read. Serious defects were revealed in the organisation of production and in the development of new types of rectifier. The developments necessary in the period 1959-65 were reviewed and it was recommended that individual factories should specialise. The main directions of technical development in rectifiers were pointed out. There are no figures, no references.

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2/2

USCOMA-DC-61,750

S/110/60/000/010/014/014  
R073/E535

**AUTHORS:** Dolin, N.F. and Zyabkina, Z.F., Engineers

**TITLE:** Conference on Transformer Construction

**PERIODICAL:** Vestnik elektropromyshlennosti, 1960, No.10, pp.77-78

**TEXT:** The conference was held in Zaporozh'ye from April 5-9, 1960. There were over 300 participants. 38 papers and 23 communications were presented. The representative of Gosplan, USSR, P. P. Chusov presented a paper on the tasks and prospects of development in the field of transformer design during 1960-1965, a period during which the production of transformers is to increase by a factor of 3.3. I. S. Kalinichenko, Elektrotechnicheskiy nauchno-issledovatel'skiy institut (Electrotechnical Scientific Research Institute), Zaporozh'ye presented a paper on the present state and prospects of development in the field of large power transformers and auto-transformers. Representatives of other works presented papers on new series and on trends of development in the transformer field. Doctor of Technical Sciences G. N. Petrov discussed the prospects of introducing multi-winding auto-transformers to satisfy the increasing requirements of inter-connected networks with differing voltages. K.K. Chernev, Glavnyy Card 1/3 ✓

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E073/E535

Conference on Transformer Construction


energeticheskii upravleniya (Chief Power Directorate), Gosplan, USSR, presented a paper in which failures of transformers produced in 1950 were analysed. I. D. Yanshina (MTZ) and B. A. Alekseyev (VNIIE) dealt with methods of investigating and monitoring the humidity of insulation. V. S. Panasyuk (MTZ) dealt with methods of drying transformers. B. V. Vanin and N. B. Leonidova (VNIIE) analysed data on the humidity of transformers in operation and on investigation of processes of drying transformer insulation. V. V. Girshberg (VNIIE) dealt with the present state and the trends of development in manufacturing complete transformer sub-stations. The problem of using aluminium in transformers was dealt with by M. A. Bass (ZTZ). A. N. Kalyakina (VEI) presented a paper on new insulation materials for transformers. Results of testing and investigation of the dynamic strength of aluminium windings of transformers with textured steel cores were presented by I. S. Nayashkov. L. V. Zamyatina and A. V. Sapozhnikov presented results of investigations of internal protection of transformers and auto-transformers from atmospheric over-voltages.

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E073/E535

Conference on Transformer Construction

Yu. A. Vasil'chenko (ZTZ) reported on methods of improving the cooling of transformers. V. I. Nepomnyashchiy, Komitet standartov, mer i izmeritel'nykh priborov Sovet Ministrov SSSR (Committee of Standards, Gauges and Metering Instruments, Council of Ministers, USSR) presented a paper on the tasks and advantages of standardization in transformer construction.



Card 3/3

DOLIN. P. A.

Collection of rules in force regarding safety methods. Moskva, Gos. energ.  
izd-vo, 1951.



DOLIN, P. A.

The repair of transformers in municipal electrical systems. Moskva, Izd-vo Ministerstva  
Kommunal'nogo khoziaistva RSFSR, 1951. 166 p. (51-34472)

TK2551.D56

DOLIN, P. A.

Pravila bezopasnosti na liniakh elektropredachi bez sniatia napriazheniia (s poiasneniiami) (Safety rules on electric transmission lines without cutting off the current (with explanations) ). M'skva, Gosenergoizdat, 1954. 63 p.

SO: Monthly List of Russian Accessions, Vol. 7, No. 6, Sep. 1954

DOLIN, P.A.

Pravila bezopasnosti pri rabotakh na liniakh elektroperedachi bez sniatia napriazheniia (s poiasneniiami); posobie dlia personala setevykh ralonov (Safety rules for work on electric power lines without interrupting the current (with explanations); manual for power line network personnel). Moskva, Gosenergoizdat, 1955. 63 p.

SO: Monthly List of Russian Accessions, Vol 7, No 9, Dec 1954

DOLIN, P.A.; KHAVIN, M.Z.; SHUTOV, I.G.; SKVORTSOV, I.M., tekhnicheskiy  
Faktor.

[Collection of safety engineering rules now in force] Sbornik  
deistvuyushchikh pravil po tekhnike bezopasnosti. Iss. 2-3  
peresm. i dop. Sostavili P.A. Dolin, M.Z. Khavin, I.G. Shutov.  
Moskva, Gos. energ. izd-vo 1955. 696 p. (MLBA 8:8)

1. Russia (1923- U.S.S.R.) Ministerstvo elektrostantsii.  
(Safety engineering)

DOLIN, PETR ALEKSEYEVICH

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622.5  
.D61

Spravochnik po tekhnike bezopasnosti Reference book on accident  
prevention techniques Moskva, Gosenergoizdat, 1958.

303 1 p. illus., diagrs., tables.

Bibliography: p. 302-304

0 8 0 1 0 5 7 11  
BEN YAMINSON, V.Ye., inzh.; DOLIN, P.A., kand. tekhn. nauk; LYASHENKO, V.D.,  
inzh.

New high-power laboratory for testing high-voltage apparatus. Elek.  
sta. 29 no.2:67-70 F '58. (MIRA 11:3)  
(Electric Laboratories)

DOLIN, P.A., kand. tekhn. nauk.

Effect of the height of the line on ice load. Elektr. sta. 29 no. 4:  
52-54 Ap '58. (MIRA 11:8)

(Electric lines---Overhead)

DOLIN, P.A., kand.tekhn.nauk

Performance of a split phase of a 400 kv. line under icing  
conditions. Elek.sta. 29 no.8:63-66 Ag '58, (MIRA 11:11)  
(Electric lines—Cold weather conditions)



DOLIN, Petr Alekseyevich; SOLOV'YEV, Petr Fedorovich, inzh.; SMIRNOV,  
A.D., inzh., obshchiy red.; BRANDENBURGSKAYA, E.Ya., red.;  
LARIONOV, G.Ye., tekhn.red.

[Safety engineering] Tekhnika bezopasnosti. Izd.3., perer. i  
dop. Moskva, Gos.energ.isd-vo.(Spravochnik elektromonters, no.5)  
(MIRA 13:4)

(Electric engineering--Safety measures)

NAYFEL'D, Mark Romanovich; DOLIN, P.A., kand.tekhn.nauk, retsenzent;  
USTINOVA, Yu.P., red.; LARIONOV, G.Ye., tekhn.red.

[Protective grounding of electric systems] Zashchitnye za-  
zemleniia v elektrotekhnicheskikh ustanovkakh. Izd.2., perer.  
Moskva, Gos.energ.isd-vo, 1959. 214 p. (MIRA 13:5)  
(Electric currents--Grounding)  
(Electric power distribution)

TUMANOV, Boris Vladimirovich; DOLIN, P.A., red.; VORONIN, K.F.,  
tekhn.red.

[Safety information for electric welders] Pamiatka po tekhnike  
bezopasnosti dlia elektrosvarshchikov. Moskva, Gos.energ.izd-vo,  
1960. 19 p. (MIRA 14:6)  
(Electric welding—Safety measures)

DOLIN, Petr Alekseyevich; BRANDENBURGSKAYA, E.Ya., red.; BORUNOV, N.I.,  
tekhn. red.

[Manual on safety engineering] Spravochnik po tekhnike bezopasnosti. Izd. 2., perer. Moskva, Gos. energ. izd-vo, 1960.  
367 p. (MIRA 14:6)  
(Industrial safety)

DOLIN, P.A.; KHAVIN, N.Z.; SHUTOV, I.G.; VORONIN, K.P., tekhn. red.

[Collection of safety engineering regulations] Sbornik deistvuiushchikh pravil po tekhnike bezopasnosti. Izd.3., peresm. i dop. Moskva, Gos.energ.izd-vo. Vol.1. 1961. 751 p. (MIRA 14:12)  
(Electric power plants---Safety measures)  
(Electric power distribution---Safety measures)

DOLIN, P.A.; KHAVIN, N.Z.; SHUTOV, I.G.; VORONIN, K.P., tekhn.red.

[Collection of regulations in industrial hygiene] Sbornik  
deistvuiushchikh pravil po tekhnike bezopasnosti. Izd.3.,  
peresm. i dop. Moskva, Gos.energ.izd-vo. Vol.2. 1962. 480 p.  
(MIRA 15:5)

(Industrial hygiene—Laws and legislation)

DOLIN, Petr ~~Alekseyevich~~; MIRONOV, P.M., red.; ORLOVA, V.Ya., red.  
izd-va; SHIL'CVSKAYA, I.Yu., red. izd-va; ISLENT'YEVA, P.G.,  
tekhn. red.

[Technical norms; a handbook] Tekhnicheskie normativy; spra-  
vochnik. Pod red. P.M.Mironova. Moskva, Metallurgizdat, 1963.  
(MIRA 16:7)

(Industrial safety--Standards)  
(Industrial hygiene--Standards)

DOLIN, P.A.; BARG, I.G., inzh., red.; BUL'DYAYEV, N.A., tekhn. red.

[Work on operating overhead electric power transmission lines; clarifications for chapter IV of "Safety Engineering Regulations for Operating Overhead Power Transmission Lines with Voltages in Excess of 1,000 volts."] Raboty pod napriazheniem na vozdushnykh liniakh elektropredachi; poiasnenia k glave IV "Pravil tekhniki bezopasnosti pri ekspluatatsii vozdushnykh lini elektropredachi napriazheniem vyshe 1 000 v." Izd.2., perer. Moskva, Gosenergoizdat, 1962. 109 p.  
(MIRA 15:9)

(Electric lines--Overhead)

(Electric power distribution--Safety regulations)



SEMENOV, N.N.; FRUMKIN, A.N.; DOLIN, P.I.

Natal'ia Alekseevna Bakh, 1895-; on her seventieth birthday  
and forty-fifth anniversary of scientific activity. Zhur.  
fiz. khim. '9 no.6:1538-1539 Je '65.

(MIRA 18:11)

DOLIN, P. I.

PROCESSES AND PROPERTIES INDEX

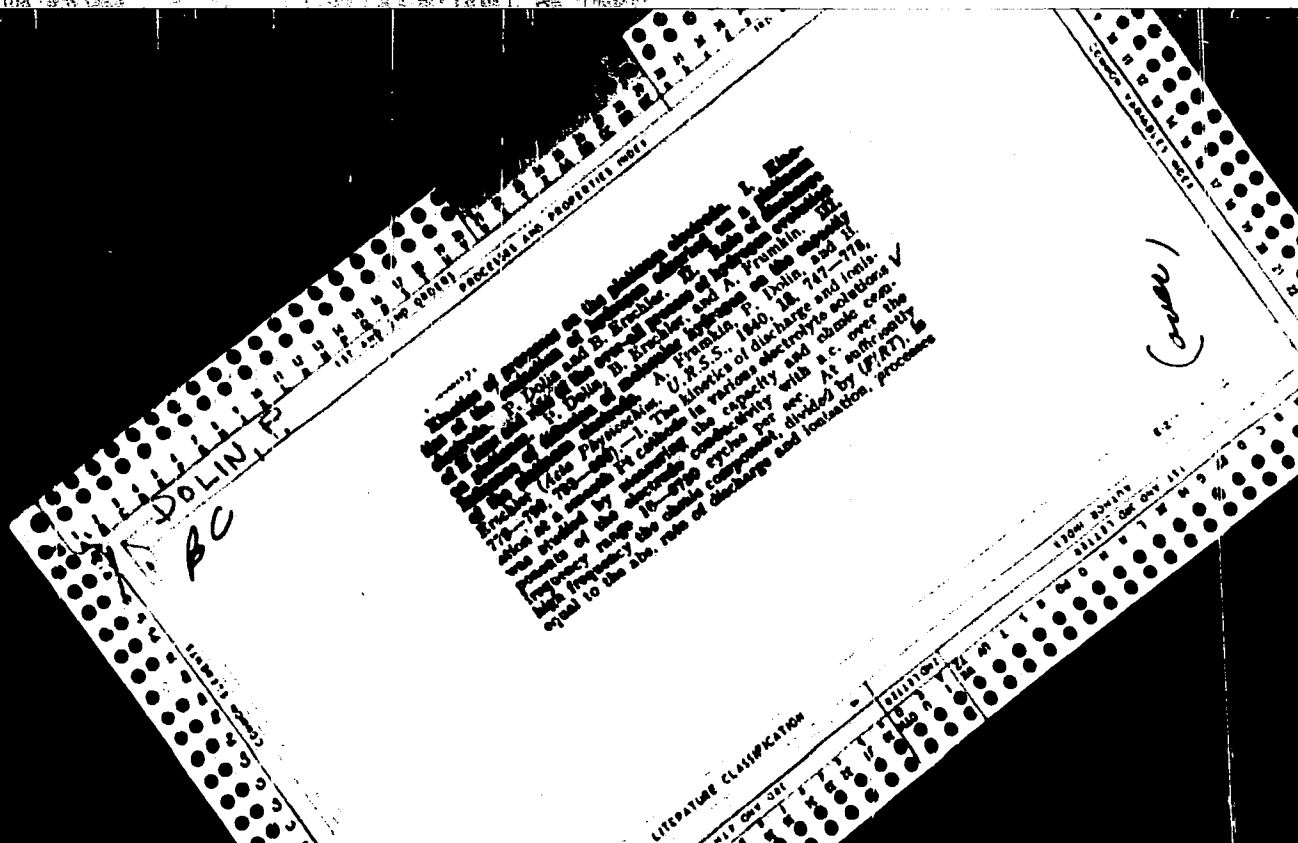
Fractionation of petroleum gases, cracked at high temperatures, with activated carbon. V. I. Shvenberg, P. I. Dolin, B. S. Zolotarev, N. P. Okhapkin and L. I. Nikolova. *Soviet. Khimich. No. 4, 13 23(1935).*

The activated C used in the fractional adsorption of constituents of the cracked gas was of Russian origin (characteristics not given). In a preliminary study of the fractions after passage through the adsorbent (mainly C), the first appearance of the desired components of the fractions after passage through the adsorbent (mainly C) was detected by means of an interferometer (data are given). The desorption was effected with superheated steam (120°). After establishing the approximate composition of the cracked gas, it was passed through a series of C adsorbents, one adsorbent for each type of gaseous hydrocarbon. Thus it was possible to obtain a divinyl concentrate of 21-22%, a propylene fraction of 81%, and an ethylene fraction of about 81%. The C when protected from the action of the higher olefins retains its activity and can be reused for adsorbing the C<sub>2</sub>, C<sub>3</sub> and C<sub>4</sub> fractions.

A. A. Borzhinsk

22

ASS. S.S.A. METALLURGICAL LITERATURE CLASSIFICATION

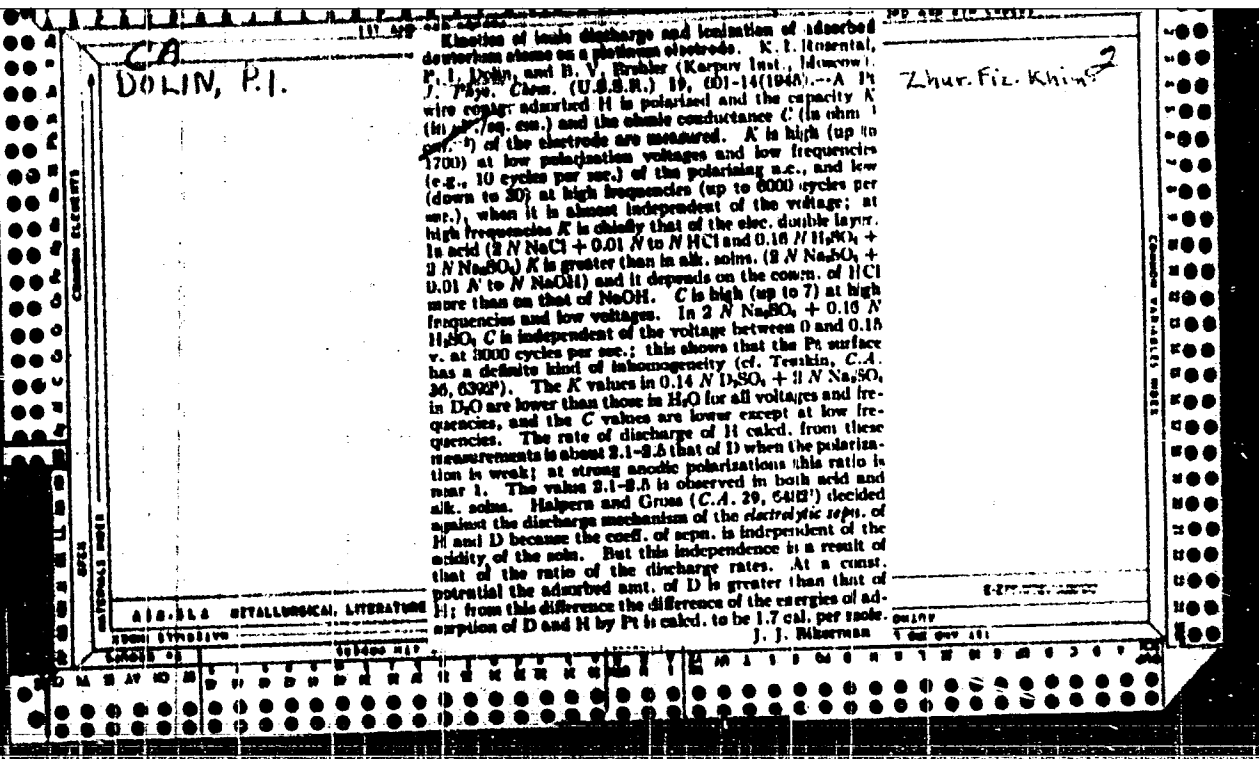


DOLIN, P.		A-1	
BC			
<p>Kinetics of processes on the platinum electrode. I. Kinetics of the ionization of hydrogen adsorbed on a platinum electrode. P. Dolin and H. Krachler. II. Rate of discharge of H ions and rate of the over-all process of hydrogen evolution on platinum. P. Dolin, H. Krachler, and A. Frankin. III. Kinetics of evolution of molecular hydrogen on the surface of the platinum electrode. A. Frankin, P. Dolin, and H. Krachler (<i>Acta Physicochim. U.R.S.S.</i>, 1940, 12, 747-778, 779-793, 793-808).—I. The kinetics of discharge and ionization at a smooth Pt cathode in various electrolyte solutions was studied by measuring the capacity and ohmic components of the electrode conductivity with a.c. over the frequency range 10-6780 cycles per sec. At sufficiently high frequency the ohmic component, divided by <math>(U/RT)</math>, is equal to the abs. rate of discharge and ionization, processes</p>			
<p>ASD-5LA METALLURGICAL LITERATURE CLASSIFICATION</p>			
<p>100-11000000</p>			
<p>100-11000000</p>			

which are thus shown to occur with fairly uniformity. The rate is independent of potential in the region where the electrode surface is covered to a not high degree, a fact accounted for by assuming that the surface is heterogeneous. The rate of discharge increases in the order  $\text{NaOH} < \text{H}_2\text{SO}_4 < \text{HCl} < \text{HNO}_3$ , which is explained by the influence of adsorption of the ions. The oxidation of Pt is very slow in acid and considerably faster in alkaline solutions.

II. By measuring the overvoltage ( $\eta$ ) of H on Pt in various electrolytes over a range of concentrations the rate of the discharge stage is found to be in HCl 27 times and in NaOH 11 times that of the over-all process  $2\text{H}^+ \rightarrow \text{H}_2$  at the reversible H potential. It is inferred that  $\eta$  on Pt is determined (a) by the rate of discharge of H ions on free and covered regions of the electrode with formation of adsorbed atoms and of  $\text{H}_2$ , respectively, and (b) by the rate of the inverse reactions. An equation relating  $\eta$  to the c.d. for a heterogeneous surface and for small values of  $\eta$  is derived.

III. Frumkin's equation (cf. A., 1963, ii, 707) for the polarization capacity of a Hg electrode is not directly applicable to Pt, since it assumes the instantaneous establishment of equilibrium between  $\text{H}^+$  in the solution and H adsorbed on the surface. Expressions for the variation of the capacity and of the ohmic component of the conductivity of a Pt electrode with frequency are derived, allowing for the finite rate of evolution of  $\text{H}_2$ .



DOLIN, P. I.

Kinetics of deuterium ions discharge and ionization of adsorbed deuterium atoms on a platinum electrode. A. Kozental, P. Dolin, and B. Ershler (Acta Physicochim. USSR, 1946, 21, 213, 234; A., 1941, 1, 338). The capacity and ohmic components of the a.c. conductivity of a Pt. electrode in acid and alkaline solutions of 2N  $\text{Na}_2\text{SO}_4$  in  $\text{H}_2\text{O}$  and  $\text{D}_2\text{O}$  have been investigated between 10 and 6000 cycles and over a range of overpotential from 0 to 1V. Charging curves of Pt were obtained in the same solutions. At potentials close to those of the corresponding gas electrodes acid solutions, and 1:2.1-2.5 in alkaline solutions. The difference in the discharge rates may be due to a greater adsorption of D than of H on Pt. The difference between the energies of adsorption of D and H on Pt is calc. from the charging curves to be 1.7 kg.-cal. This indicates that the difference in activation energies for discharge of D and H is not great. The capacity and ohmic components of the conductivity of a Pt electrode were investigated in aq. 2N NaOH in the presence of 0.01-mole of NaOH. Results agree with the Volmer-Hrunkin theory of slow discharge. The variation in the rate of discharge of H and D with potential indicates that the surface of Pt is of a more complex inhomogeneity than was inferred on the basis of charging curves alone (cf. loc. cit.). U.D.S.

Sci. Res. Physico-Chemical Inst. im L. Ya. Karpov

DOBIN, P.I.		PROCESSES AND PROPERTIES INDEX																																																																																											
<p>Measurement of the polarization capacity of the mercury electrode in a wide frequency range. V. I. Melnik, G. A. Zayen and P. I. Dolin. Doklady Akad. Nauk S.S.S.R. 66, 400-12 (1940). The electrode capacity <math>C</math> was defd. as a function of the potential <math>E</math> with the aid of a perfected app. permitting measurements up to 200,000 hertz without sacrifice of accuracy (<math>\pm 1\%</math>) and up to 500,000 hertz with a somewhat lower accuracy. In a pure 1 N KCl soln., <math>C</math> at a given <math>E</math> proved to be independent of the frequency in the whole range, hence, the time necessary for the establishment of the double layer is certainly shorter than <math>10^{-6}</math> sec. In the same soln. satd. with hexanol, <math>C</math> varies with the frequency. With increasing frequency, the height of the peak bordering on the range of low <math>C</math> and corresponding to adsorption decreases, and disappears altogether at a sufficiently high frequency, the lower the smaller the concn. of the surface-active substance and the greater its adsorbability. This frequency permits estn. of the time necessary for the establishment of the adsorption equil. Higher frequency also results in some lowering of <math>C</math> in the polarization range where <math>C</math> does not vary with <math>E</math>, evidently owing to the slowness of penetration of the ions into the adsorbed hexanol layer on the surface of the Hg. Expts. with AuOH gave analogous results. N. Thon</p>																																																																																													
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<table border="1"> <thead> <tr> <th colspan="10">GROUPS</th> <th colspan="10">SUBGROUPS</th> <th colspan="10">SUBSUBGROUPS</th> </tr> </thead> <tbody> <tr> <td>1</td><td>2</td><td>3</td><td>4</td><td>5</td><td>6</td><td>7</td><td>8</td><td>9</td><td>10</td> <td>11</td><td>12</td><td>13</td><td>14</td><td>15</td><td>16</td><td>17</td><td>18</td><td>19</td><td>20</td> <td>21</td><td>22</td><td>23</td><td>24</td><td>25</td><td>26</td><td>27</td><td>28</td><td>29</td><td>30</td> </tr> <tr> <td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td> <td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td> <td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td> </tr> </tbody> </table>				GROUPS										SUBGROUPS										SUBSUBGROUPS										1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30																														
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DOLIN, P.I.; LOSEV, V.V.; LUKOVITSEV, P.D.; MEDVEDOVSKIY, P.D.

Letter to the editor. Usp. khim., 22 no.6:775-776 Je '53. (MLRA 6:5)  
(Electrochemistry--History)

*DOLIN*

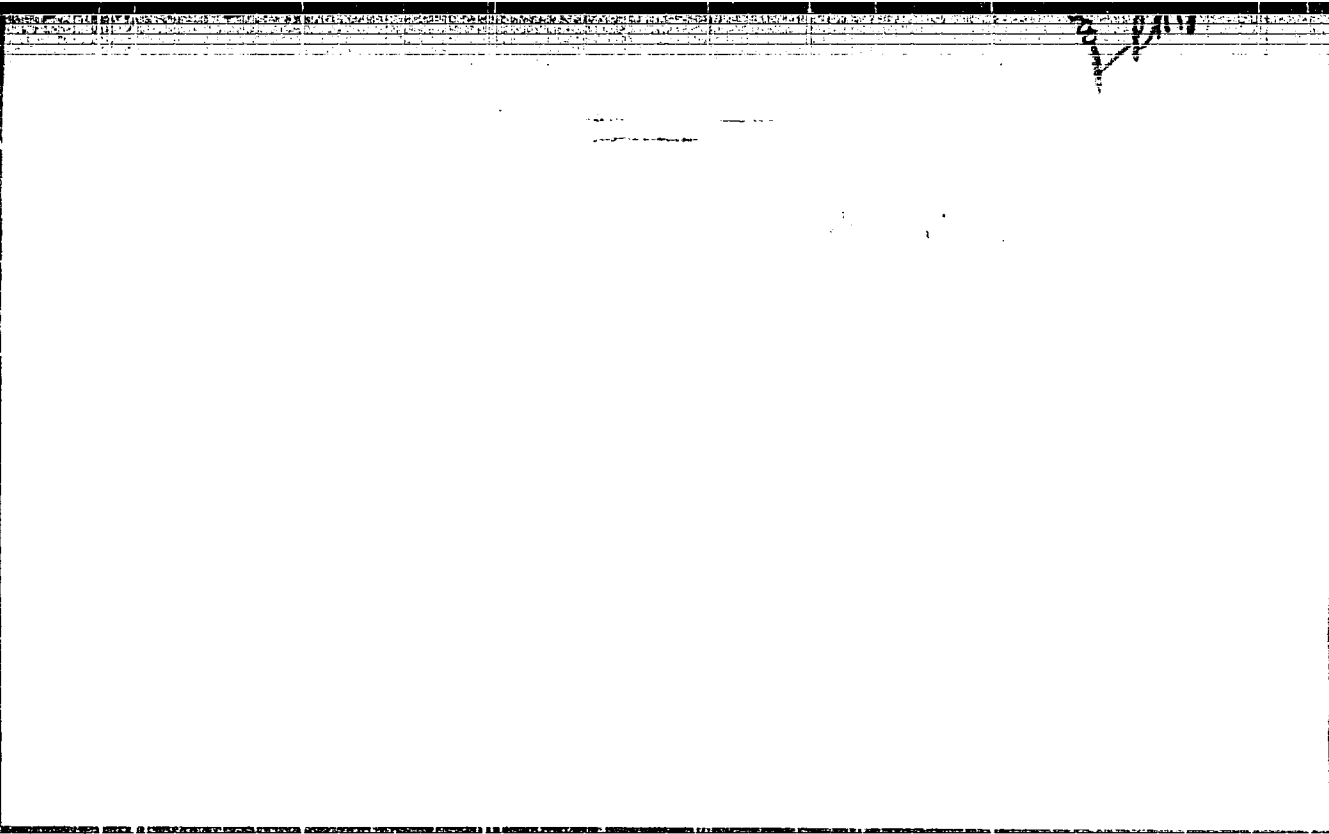
BAKH, N.A., professor, doktor khimicheskikh nauk, redaktor; VERESHCHINSKIY, I.V., redaktor; DOLIN, P.I., redaktor; MYASHNIKOV, I.A., redaktor; KISILEVA, A.A., ~~tekhnicheskii~~ redaktor.

[Collection of papers on radiation chemistry] Sbornik rabot po radiatsionnoi khimii. Moskva, 1955. 262 p. (MLRA 8:11)

1. Akademiya nauk SSSR.  
(Radiation)

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000410820010-5



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*120114, P. 2*

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*1*

Influence of dissolved oxygen on the radiolysis of water by the action of  $\gamma$ -rays. P. I. Dolig. *Sbornik Rabot Radiatsionnoi Khim., Akad. Nauk SSSR*, 1955, 7-23. Doubly distd. water was exposed to  $\gamma$ -radiation from a H<sub>2</sub>-gen tube having a cylindrical W anode, 30 mm. in diam., at 65 kv., and 200 mm., anode current, or to  $\gamma$ -radiation from a Co<sup>60</sup> source. For dosages of  $0.15 \times 10^{21}$  e.v./ml., H (formed by the reaction  $2H_2O \rightarrow H_2 + H_2O_2$ ) varies linearly with dosage in O-satd. water, if the gaseous reaction products are removed. If the gaseous products are permitted to build up pressure, a steady state is reached beyond which further changes in the pressure of the H,  $P_{H_2}$ , and the concn. of H<sub>2</sub>O<sub>2</sub>,  $C_{H_2O_2}$ , are negligible. In O-satd. H<sub>2</sub>O, this  $P_{H_2}$  is 226 mm. Hg. and  $C_{H_2O_2} = 0.93 \times 10^{-3} M$ . These pressures and concns. are reduced when pressure of the O in the system is reduced. The presence of H<sub>2</sub>SO<sub>4</sub> at concns. of 0.1 to 1N increases radiolysis in conformity with the kinetic mechanism based on the ionic disocn. of the free radicals HO<sub>2</sub> and OH. KOH increases radiolysis at concns. as low as  $10^{-3} N$ . C. H. Fuchman

*Chem*

*RM*

DOLIN, P. I.

Kinetics of the reaction of hydrogen peroxide with hydrogen gas in aqueous solution under the influence of x-rays.  
P. I. Dolin. *Sbornik Rabot Radiatsionnoi Khim., Akad. Nauk SSSR*, 1955; 21-35. -- Studies of the kinetics of radiation-induced decompu. of  $H_2O_2$  by  $H_2$  indicate the existence of 4  $H_2O_2$  concn. ranges characterized by different functional relations of decompu. rate with concn. The concn. ranges for regions I-IV, resp., are: 0.0005-0.0015, 0.0015-0.012, 0.002-0.01, 0.01-1.15 moles/l. The regions are characterized by the following resp. values for  $-d[H_2O_2]/d(tJ)$  where  $J$  is the dosage power: region I,  $-K_1 + (K_1/[H_2O_2])$ , where  $K_1$  is  $0.34 \times 10^{-4}$  moles./c.v.;  $K_2 = 1.3 \times 10^{-4}$  (moles.)<sup>2</sup>/c.v.; region II,  $4.20$  moles./100 e.v.; region III,  $K_1 + K_2 [H_2O_2]$ , where  $K_1$  is  $1.5$  moles./c.v.,  $K_2 = 1.8 \times 10^{-4}$  moles./100 e.v. (moles); region IV,  $K_1 + K_2 [H_2O_2]$ , where  $K_1$  is  $17$  moles./c.v. and  $K_2$  is  $60$  moles./100 e.v. (moles)<sup>2</sup>/c.v.  
G. H. Fuchsman

Chem Phys

PM 8/24

*DOLIN, P.I.*

USSR/Scientists - Chemistry

Card 1/1      Pub. 147 - 1/21

**Authors** : Rebinder, P. A.; Dolin, P. I.; Kabanov, B. N.

**Title** : The work of A. N. Frumkin and his school in the field of surface phenomena and kinetics of electron processes

**Periodical** : Zhur. fiz. khim. 29/10, 1746-1750, Oct 1955

**Abstract** : Honoring the 60-th birthday of the famous Soviet physico-chemist, Academician Aleksandr Naumovich Frumkin, a group of his colleagues published a list of Frumkin's scientific research work on surface phenomena and the kinetics of electrode processes.

**Institution** : .....

**Submitted** : .....

DUZHENKOV, V. I. and DOLIN, P. I.

"The Influence of X-rays on Dilute Water Solutions of Organic Substances" p.96

Trudy Transactions of the First Conference on Radioaction Chemistry, Moscow,  
Izd-vo AN SSSR, 1958. 330pp.  
Conference 25-30 March 1957, Moscow

BRUSENTSEVA, S. A., DOLIN, P. I.

"The Influence of Potassium Bromide and Potassium Chloride Concentration on the Yield of Molecular Products From Radiolysis of Water Solutions" p.40

Trudy Transactions of the First Conference on Radioaction Chemistry, Moscow,  
Izd-vo AN SSSR, 1958. 330pp.  
Conference -25-30 March 1957, Moscow



BAKH, N.A., prof., otv. red.; MEDVEDEV, S.S., red.; VESELOVSKIY, V.I., prof., red.  
DOLIN, P.I., doktor khim. nauk, red.; MILLER, M.B., kand. khim. nauk, red.  
TSIMLIN, B.L., kand. khim. nauk, red.; TRIFONOV, D.N., red. izd-va.;  
BUGAYENKO, L.T., red. izd-va.; MOSKVICHENKO, N.I., tekhn. red.

[All-Union Conference on Radiation Chemistry; transactions] Trudy I  
Vsesoiuznogo sobeshchaniia po radiatsionnoi khimii. [Moscow, 1957.]  
Moskva, Izd-vo Akad. nauk SSSR, 1958. 330 o. (MIRA 11:11)

1. Chlen-korrespondent AN SSSR (for Medvedev).  
(Radiochemistry)

DOLIN, P.I

p 2

PHASE I BOOK EXPLOITATION

790

Akademiya nauk SSSR. Otdeleniye khimicheskikh nauk

Deystviye ioniziruyushchikh izlucheniya na neorganicheskiye i organicheskiye sistemy (Effect of Ionizing Radiation on Inorganic and Organic Systems)  
Moscow. Izd-vo AN SSSR, 1958. 416 p. 7,000 copies printed.

Resp. Ed.: Pshezhetskiy, S. Ya.; Ed. of Publishing House: Bugayenko, L.T.;  
Tech. Ed.: Frusakova, T. A.

PURPOSE: This publication is for scientists working in the field of radiochemistry.

COVERAGE: This collection of articles represents contributions of Soviet scientists in the field of radiochemistry. The papers are concerned with the effect of ionizing radiation on organic and inorganic substances in solutions and in the solid phase. These papers were completed in the years 1951 - 1956 at the Institute of Physical Chemistry, AS USSR, the Institute of Physics and Chemistry imeni L. Ya. Karpov, the Moscow State University, and other scientific institutions. Most of these works are a continuation of those published in "Sbornik rabot po radiatsionnoy khimii" published in 1955. Ts. I. Zalkind and Yu. M. Malinskiy cooperated in the editing of this symposium.

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- Effect of Ionizing Radiation (Cont.)

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- TABLE OF CONTENTS:

PART 1. REACTIONS IN AQUEOUS SOLUTIONS  
AND RADIATION AND ELECTRO-CHEMICAL PROCESSES

Preface

Duzhenkov, V.I., Dolin, P.I. Effect of X-ray Irradiation on Aqueous Alkali Solutions Saturated With Oxygen

7

The kinetics of accumulation of molecular products formed in the radiolysis of water are studied in this paper. These products are: hydrogen peroxide and hydrogen. The absorption of oxygen in high-purity alkali solutions saturated with oxygen was also taken into consideration. It was determined that the initial yield of hydrogen depends on the concentration of the irradiated KOH solution only for concentrations up to 0.6 - 0.7 N KOH. The same relation was found for  $H_2O_2$ . The material balance of the molecular products showed a strong deviation towards excessive absorption of oxygen. This fact was explained as the formation of higher peroxides, probably  $HO_2$  or the complex  $H_2O_2 \cdot HO_2$ .

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Effect of Ionizing Radiation (Cont.)

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There are 4 figures and 8 references, of which 6 are Soviet and 2 English.

Gvozdev, B.A., Shubin, V.N. Effect of Accelerated Electrons on Potassium Permanganate Solutions

12

The reduction of aqueous  $\text{KMnO}_4$  solutions by accelerated electrons in the concentration range of  $10^{-3}$  to  $3 \cdot 10^{-1}$  M is discussed in this paper. The yield of the reaction  $\text{Mn(VII)} \longrightarrow \text{Mn(IV)}$  is determined from the relation between the amount of reduced permanganate and the amount of energy absorbed by the solution. The authors explain the effect of the acidity of the solution (in the range pH = 0.4 to 12) on the rate of reduction. The maximum yield was observed for pH = 2.05. It was determined that the yield of reaction does not depend on the dosage rate in the range from  $\sim 10^{17}$  to  $\sim 10^{20}$  eV/ml-sec for electron energies from 0.1 to 0.7 Mev. There are 8 figures, 3 tables, and 7 references, of which 4 are Soviet, 2 English, and 1 Polish.

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Effect of Ionizing Radiation (Cont.)

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Sidorova, L.P., Zimin, A.V., Proskurin, M.A. Effect of  $Co^{60}$   $\gamma$ -Radiation

22

on Aqueous Solutions of Salts of Tin

The article deals with the effect of irradiation on  $SnCl_2$  and  $SnCl_4$  solutions, and the role of the salts as acceptors of the products of water radiolysis. Experimental data show that the irradiation of  $1 \cdot 10^{-2}$  M solutions of  $Sn^{2+}$  in 4N HCl and in 1N NaOH results in a hydrogen yield equal to the yield of bivalent tin oxidation. The yield of hydrogen is 0.53 - 0.60 molecules/100 ev of absorbed energy. The hydrogen yield does not change with the increase of dosage in an acid medium, and decreases in an alkaline medium. The presence of bivalent tin in alkaline solutions increases the hydrogen yield, while it does not show an essential effect in an acid medium. The oxidation of  $Sn^{2+} \rightarrow Sn^{4+}$  in acid and alkaline media is apparently conditioned by the interaction of  $Sn^{2+}$  with  $H_2O_2$  molecules. There are 3 tables, 1 figure, and 14 references, of which 6 are Soviet, and 8 English.

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Effect of Ionizing Radiation (Cont.)

790

Chernova, A.I., Orekhov, V.D., Proskurin, M.A. Oxygen Compounds of Iron Ions and Their Conversion During Radiolysis of Aqueous Solutions

29

This is a study of iron peroxide compounds and their behavior under  $\gamma$ - and ultraviolet irradiation. Variable-valence ions are considered to be carriers of the oxidizing components in radiochemical oxidation. The peroxide iron complex which forms during the irradiation of a Mohr salt solution in 4N sulfuric acid decomposes in the presence of  $Fe^{3+}$  to form bivalent ions. This process is intensified with the increase of the  $Fe^{3+}$  concentration in the solution. It was determined that the peroxide iron complex can be formed also in Mohr salt solutions in 0.8N acid when an excess of  $Fe^{3+}$  is present. Addition of  $MnSO_4$ ,  $TiNO_3$ ,  $NaCl$ ,  $KBr$ ,  $KI$ ,  $Ce(NO_3)_3$  and  $CuSO_4$  to the irradiated solutions of the Mohr salt in 0.8 and 4N sulfuric acid leads to an increased yield of  $Fe^{3+}$  ions. The effect is due to the presence of bromide and iodide ions as well as  $Ce^{3+}$  and  $Cu^{2+}$ . There are 6 figures and 11 references, of which 4 are Soviet, 4 English, and 3 German.

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Effect of Ionizing Radiation (Cont.)

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Sharpatyy, V.A., Orekhov, V.D., Proskurin, M.A. Sensitization of the Radiolytic Conversion of Sodium Nitrate in Aqueous Alkaline Solutions

37

The subject of this paper is the effect of the temperature of the solution on the yield of radiolytic conversion of nitrate in aqueous alkaline solutions at temperatures from 20° to 90°. The same process was studied with glycerin as acceptor of OH radicals. An increase from 20° to 40° in 1M NaNO<sub>3</sub> / 1M KOH causes a sharp increase of the nitrate yield: from ~ 3.0 to ~ 6.5 equiv./100ev. The increase in yield reaches its limit value at 80° and equals about 8 equiv./100ev. The sensitizing effect of glycerin is apparent only when its concentration is ~ 5.10<sup>-4</sup> M and remains constant for concentrations up to 10<sup>-1</sup> M. The presence of molecular oxygen (air) inhibits this effect. There are 5 figures and 10 references, of which 6 are Soviet and 4 English.

Sharpatyy, V.A., Orekhov, V.D., Proskurin, M.A. Radiolytic Reduction of Sodium Nitrate in Concentrated Aqueous Solutions

43

This paper considers the radiolytic reduction of nitrate solutions in a wide range of concentrations. Concentrations of NaNO<sub>3</sub> above 1M in the presence of an inert gas (nitrogen) resulted in a yield of 8 to 9 equiv./100ev NO<sub>2</sub><sup>-</sup>. It was found that molecular oxygen inhibits the reduction

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## Effect of Ionizing Radiation (Cont.)

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process, which is evident in lower results as compared to the process in an inert atmosphere. There are 2 figures and 6 references of which 3 are Soviet and 3 English.

Chernykh, V.Ya., Pshezhetskiy, S.Ya., Tyurikov, G.S. Kinetics of the Decomposition of Hydrogen Peroxide Under the Action of  $\gamma$  - Radiation

48

The authors studied the kinetics of  $H_2O_2$  decomposition in a wide range of concentrations: 1.78 - 92.2 mol.%, in  $\gamma$  and ultraviolet radiation. The rate of the thermal reaction was also taken into consideration. The rate of the radiation reaction is proportional to the square root of  $\gamma$ -radiation intensity. The activation energy equals  $6.5 \pm 1.0$  kcal/M. The activation energy of the photochemical reaction is 8 - 9 kcal/M. The efficiency of the  $\gamma$ -radiation reaction is 21 (at  $-4^\circ$ ) to 230 molecules of  $H_2O_2$  (at  $50^\circ$ ) per 100ev dosage rate of  $1.84 \cdot 10^{18}$  ev/l. sec. It is a chain reaction. The kinetics of the radiation, photo, and thermal reactions can be clarified by assuming electrolytic dissociation of the  $HO_2$  radical, the rate of which depends on the concentration of the solution. There are 20 figures, 4 tables, and

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Effect of Ionizing Radiation (Cont.)

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22 references of which 3 are Soviet, 16 English, and 3 German.

Zalkind, Ts.I., Veselovskiy, V.I. Mechanism of Radiochemical Formation of Stationary Potential Differences in Aqueous Solutions

66

The stationary potential difference of  $\sim 0.9$  is formed in the system  $\text{Pt}/\text{H}_2\text{SO}_4$  saturated with nitrogen/Au and irradiated with  $\text{Co}^{60}$   $\gamma$ -radiation. It was shown that the formation of a positive potential at the Au electrode is connected with the radiolytic formation of the OH radical. The oxidation of the electrode during heating facilitates the formation of the positive potential at the Au electrode. The rate of reduction is determined by the rate of the electrode reaction, i.e., the electrochemical discharge stage. There are 6 figures, and 6 references of which 5 are Soviet and 1 English.

Zalkind, Ts.I., Veselovskiy, V.I. Photoelectrochemical and Radiation Electrochemical Processes in Aqueous Solutions of Uranium Salts

74

Uranium salts were irradiated with radon and  $\text{Co}^{60}$ . It was shown that the hexavalent uranium salts show reduction of uranyl ions to pentavalent uranium ions. Due to the ease of oxidation - reduction transitions in the system  $\text{U(VI)} / \text{U(V)}$ , the increase of the uranyl ion concentration is followed by a decrease in the amount

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of the formed  $H_2O_2$ , uranium peroxide compounds, and in the oxidation of U(IV) and oxalic acid. Irradiation of the system U(IV) / U(VI) and U(III) / U(IV) results in a shift of equilibrium and the formation of more oxidized forms. There are 14 figures, 7 tables, and 18 references of which 8 are Soviet, 7 English, and 3 German.

Miller, I.B., Veselovskiy, V.I. Radiation Electrochemical Processes in Aqueous Solutions of Uranyl Salts

93

This is a study of the electrochemical nature of the redox components in the radiolysis of uranyl salt solutions. Certain conditions were established for the formation of the "hydrogen" and "oxygen" potentials in this system. A stationary potential of the Pt electrode develops during  $\gamma$ -irradiation due to the emergence of nonequilibrium concentrations of U(V). The stationary potential at the Au electrode in uranyl sulfate solutions is  $\sim 1.1V$ , while at the Pt electrode it shifts towards negative values. There are 9 figures and 12 references, of which 6 are Soviet and 6 English.

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Rozenfeld, I.L., Oshe, Ye.K. Mechanism of Activation of Electrodes of Local Cells During Irradiation

103

This paper discusses the effect of radiation on the properties of semiconducting oxide films of the electrodes. It was determined that only irradiation of the cathode intensifies the corrosion of metals in electrolytes. This is due to the sharp increase in the cathode efficiency resulting from "radiation conductivity" in the protective film which is regarded as a semiconductor. The radiation dependence of the corrosion current is expressed by:

$$I_k = I_k^0 + A\sqrt{I},$$

where A is the constant for the given pair of electrodes. Candidate of physical and mathematical sciences V.B. Sandomirskiy participated in this work. There are 10 figures and 6 references of which 1 is Soviet, 3 English, 1 French and 1 Czech.

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Effect of Ionizing Radiation (Cont.)

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Duzhenkov, V.I., Iolin, P.I. Effect of X-rays on Diluted Aqueous Solutions of Organic Substances

114

The effect of organic additives on the yield of molecular products from the radiolysis of water is studied in solutions of varying acidity. It was determined that the initial yield of  $H_2$  and  $H_2O_2$  increases, and their fixed concentrations are lowered. This appears to be due to the fact that the radical products of radiolysis:  $H$ ,  $OH$ ,  $EO_2$  react with the organic molecules more vigorously than the molecules  $H_2$  and  $H_2O_2$ . There are 3 figures, 2 tables, and 9 references, of which 3 are Soviet and 6 English.

Chernov, A.I., Orekhov, V.D., Proskurin, M.A. Sensitization and Inhibition of Radiolytic Reduction of the Uranyl Ion in Aqueous Solutions

120

The radiolytic reduction of uranyl ions was sensitized with glucose and glycerin, and inhibited with methylene blue. The reduction of  $UO_2^{2+}$  does not exceed 0.5 molecules/100 ev for low concentrations of the uranyl ion ( $5 \cdot 10^{-3}$  M) and of glucose or glycerin ( $5 \cdot 10^{-3}$  M). Concentration of 0.5 M glycerin leads to reduction of 5 molecules/100 ev. The inhibiting effect of methylene blue is modified by the ratio of concentrations of two substances reacting with  $H$ ,

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the product of radiolysis: for  $C_{\text{O}_2^{2+}} : C_{\text{dye}} = 1:1$  the dosage of inhibition of the dye reaches  $10^{18}$  ev/ml and decreases with the increase of this ratio. The participation of ionized and excited molecules of water in these processes is taken into consideration.  $\text{Co}^{60}$   $\gamma$ -radiation was used for irradiation. There are 3 figures and 19 references of which 8 are Soviet and 11 English.

PART 2. REACTIONS OF INORGANIC SUBSTANCES

Buneeyev, N.A., Myasnikov, I.A., Pshezhetskiy, Formation of Ozone in Liquid Oxygen Due to  $\gamma$ -Radiation

129

The formation of ozone in LOX due to  $\text{Co}^{60}$  radiation was studied with respect to duration and intensity of irradiation. It was found that the yield of ozone is 13.6 molecules per 100 ev of the absorbed  $\gamma$ -radiation energy. There are 4 tables, 3 figures, and 3 Soviet references.

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Effect of Ionizing Radiation (Cont.)

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Buneyev, N.A., Pshezhetskiy, S.Ya., Myasnikov, I.A. Formation of Ozone in Gaseous Oxygen Due to Fast Electrons

133

The effect of electrons on oxygen was studied at room temperature and atmospheric pressure, under dynamic and static conditions. The determined stationary concentration of ozone equals 0.1 - 0.2 per cent for 200 Kev electrons and current of 50 - 100  $\mu$ a. This is 4 to 8 times less than for LOX under analogous conditions. The efficiency of the reaction is  $\sim$  1.5 molecules per 100 ev. There are 9 figures, 5 tables, and 10 references, of which 3 are Soviet, 4 English, 2 German and 1 Belgian.

Dmitriyev, M.T., Pshezhetskiy, S.Ya. Radiation Oxidation of Nitrogen. Part 1. Kinetics of the Nitrogen Oxidation Reaction due to Electron Impact and the Effect of Ionization Processes

145

This paper deals with kinetics of ionization and oxidation of nitrogen due to the impact of 0 - 400ev electrons at pressures up to 0.1 mm Hg. The reaction rate is defined by an equation of the second order and the activation energy of the reaction equals  $7.4 \pm 1$  kcal/M. The obtained data show that ionization of molecular nitrogen is the primary elementary process leading to oxidation of nitrogen, and that it is followed by direct interaction of the molecular nitrogen ion with a molecule of oxygen and the interaction

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of oxygen with products of the dissociation of the molecular nitrogen ion. There are 18 figures, 12 tables, and 27 references of which 4 are Soviet, 19 English, 3 German, and 1 Belgian.

Dmitriyev, M.T., Pshchetskiy, S.Ya. Radiation Oxidation of Nitrogen.

171

Part 2. Kinetics of the Oxidation of Nitrogen Due to Fast Electrons

The oxidation of nitrogen due to fast electrons is studied in the gaseous phase and at atmospheric pressure. The reaction kinetics are defined by an equation of the second order. The reaction rate is proportionate to the electron flux and to the ratio of gas layer thickness to the maximum electron path in gas. At the temperature of 40° the yield is 1.3 atoms of bound nitrogen for air, and 2 atoms for a 1 : 1 gas mixture per 100 ev of absorbed energy. The differences in the absolute values of reaction rate constants are interpreted as the result of variations in the recombination of ions. There are 10 figures, 2 tables, and 9 references of which 7 are Soviet and 2 English.

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Effect of Ionizing Radiation (Cont.)

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Bol'shun, Ye.V., Pahezhet'skiy, S.Yu., Myasnikov, I.A. Formation of Hydrazine in Liquid Ammonia Due to Fast Electrons

182

The formation of hydrazine due to fast electrons leads to a stationary hydrazine concentration conditioned by the balancing of the rates of direct and inverse reactions. The reaction efficiency is 1 - 1.2 molecules of hydrazine per 100 ev. There are 2 tables, 1 figure, and 6 references of which 3 are Soviet and 3 English.

Baberkina, A.S., Froskurnin, M.A., Orekhov, V.D. The Effect of  $\gamma$ -Radiation on Solid Potassium Nitrate

186

The  $\gamma$ -irradiation of potassium nitrate leads to evolution of gas and nitrite formation. The irradiated salt shows a sharp increase of gas evolution when heated to  $\sim 129^\circ$ . A decrease in the amount of initially formed nitrite is observed when the temperature rises above  $122^\circ$  (temperature of lattice destruction for  $\text{KNO}_3$ ). The decrease in the nitrite yield is connected with the inverse reaction of nitrate formation which occurs due to the presence of atomic oxygen or free radicals in the irradiated salt.

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Effect of Ionizing Radiation (Cont.)

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There are 4 figures and 12 references of which 2 are Soviet, 9 English, and 1 German.

Baberkin, A.S., Proskurnin, M.A., Orekhov, V.D. Effect of  $\gamma$ -Radiation on Solid Potassium Perchlorate and Chlorate

193

Potassium perchlorate and chlorate powders were subjected to the effect of  $\text{Co}^{60}$   $\gamma$ -radiation in doses up to 10 megarentgens per 1 g of the salt. It was established that the conversion of  $\text{KClO}_4$  at temperatures up to 100° yielded  $\sim 1.1$  molecules of  $\text{KClO}_3$ /100 ev and no other products. At temperatures up to 200° the yield of the chlorate drops, the chloride appears and gas evolution occurs. At temperatures above 320° the yield of the chlorate drops to 0.7, and the amount of the other products increases. The irradiated  $\text{KClO}_3$  yields 0.4 molecules of  $\text{KCl}$  and 1.2 molecules of  $\text{KClO}_2$  per 100 ev. At elevated temperatures the yield of the chlorite drops to zero. It is assumed that the thermal decomposition of irradiated salts has a radical character facilitating the conversion of the intermediate products:  $\text{ClO}_4$ ,  $\text{ClO}_3$ , O, and K. Doses not exceeding 10 megarentgens per 1 g cause coloration of the salts. This is connected with the formation of coloration centers in the crystal lattice which disappear at temperatures above 100°.

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Effect of Ionizing Radiation (Cont.)

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There are 7 figures and 10 references of which 3 are Soviet, 6 English, and 1 German.

PART 3. REACTIONS OF ORGANIC SUBSTANCES

Krongauz, V.A., Bagdasar'yan, Kh.S. Energy Transfer in the Radiolysis of Benzoyl Peroxide Solutions

205

This paper considers the radiolysis of benzoyl peroxide solutions in benzene, cyclohexane, and ethyl acetate. In the benzene solution of peroxide, the transfer of excitation energy is from the solvent to the solute. This effect was not detected in cyclohexane or ethyl acetate solutions, in which the radiation decomposition of peroxide shows a chain mechanism. The authors determined yields of radicals for the solvents and the peroxide. Small additions of anthracene and phenanthrene to benzene inhibit the decomposition of peroxide. There are 7 figures, 4 tables, and 15 references of which 2 are Soviet, 11 English, and 2 German.

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Effect of Ionizing Radiation (Cont.)

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Mikhaylov, B.M., Tarasova, L.V., Bogdanov, V.S. Radiochemical Conversion of Organic Substances. Part 1. Conversion of Gaseous Aliphatic Hydrocarbons Due to Fast Electrons

218

Methane, ethane, propane, and n-butane were irradiated with a 90 Kev electron beam. Dehydrocondensation is the basic process of radiolysis. Liquid hydrocarbons constitute 50 percent of the conversion products of methane, and 70 - 90 percent of the conversion products of ethane, propane, and n-butane. There are 3 tables, 2 figures, and 7 English references.

Mikhaylov, B.M., Kumova, M.Ye., Bogdanov, V.S. Radiochemical Conversion of Organic Substances. Part 2. Oxidation of Methane with Oxygen Due to Fast Electrons

223

A mixture of methane and oxygen ( $\text{CH}_4 : \text{O}_2 = 1 : 1$  and  $1 : 1$ ) was irradiated with fast electrons.  $\text{CO}$ ,  $\text{CO}_2$ ,  $\text{H}_2$ , and  $\text{H}_2\text{O}$  were found in the reaction products. About 50 percent of the methane was converted to the liquid phase. The rate of oxidation increases with methane content (50 - 80%), and with increased pressure (190 - 760 mm Hg). The mechanism of the reaction is regarded as radical.

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Effect of Ionizing Radiation (Cont.)

790

There are 9 figures, 4 tables, and 19 references of which 8 are Soviet, 10 English, and 1 German.

Vereshchinskiy, I.V., Bakh, N.A. Oxidation of Organic Compounds with Molecular Oxygen Due to Ionizing Radiation. Part 5. Effect of Radiation From a Nuclear Reactor on Benzene and Cyclohexane in Presence of Molecular Oxygen

234

Liquid benzene and cyclohexane were subjected to radiation from a nuclear reactor (recoil protons and  $\gamma$ -radiation) in the presence and absence of oxygen. Benzene oxidation products included phenol and aldehydes. Cyclohexanes gave hydroperoxides, disubstituted peroxides, cyclohexanone, and acids. Yields are similar to those obtained from low-density ionization. Gaseous products of benzene radiolysis in vacuum contain hydrogen and acetylene. The amount of molecular hydrogen increases in the presence of oxygen. The radiolysis of cyclohexane in vacuum yields hydrogen in amounts equal to number of molecules of cyclohexane which take part in polymerization. The presence of oxygen does not increase the hydrogen yield. There are 11 figures, 1 table, and 31 references of which 8 are Soviet, and 23 English.

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Effect of Ionizing Radiation (Cont.)

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Sarayeva, V.V., Bakh, N.A., Rybin, L.V., Larin, V.A. Oxidation of Organic Compounds with Molecular Oxygen Due to Ionization Radiation. Part 6. Identification of Individual Products Obtained From the Oxidation of N-Heptane and Isooctane

248

This paper discusses the identification of individual carbonyl compounds and acids, and the accumulation of alcohols formed during the radiolytic oxidation of n-heptane and isooctane. It was shown that most of the carbonyls are composed of compounds with the number of carbons close to that of the original hydrocarbon molecule. There are 15 figures, 7 tables, and 15 references of which 7 are Soviet, 7 English and 1 French.

Sarayeva, V.V., Rudenko, B.I. Radiolytic Oxidation of Acetone

263

Acetone was irradiated with  $\text{Co}^{60}$   $\gamma$ -radiation in doses of  $2 \times 10^{13}$  and  $1 \times 10^{14}$   $\text{ev/cm}^3$ . sec and with X-rays of  $2 \times 10^{16}$   $\text{ev/cm}^3$  sec. in the presence of oxygen and at temperatures of 18 - 20°. Acids and aldehydes were the products of oxidation. No peroxides were detected. The amount of aldehydes increased with the increase of the dosage rate. The initial aldehyde yield is ~ 80 molecules per 100  $\text{ev}$  and it is independent of the dosage rate. There are 2 figures and 6 references, 3 of which are Soviet and 3 English.

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Sarayeva, V.V., Kinetics of Interaction of Organic Peroxides With the Iodide Ion in Acetic Acid and the Determination of Peroxides Obtained From Radiolytic Oxidation of Hydrocarbons

266

This is a study of the rate of interaction of sixteen organic peroxides with potassium iodide in glacial acetic acid. It was determined that the peroxides differ in rate constants but the rate constant for a given type of peroxides varies only slightly with changes in the structure of the hydrocarbon chain of the peroxide. Radiolytic oxidation yields hydroperoxides and peroxides from n-heptane, isooctane and toluene. Cyclohexane and tetralin yield only hydroperoxides. There are 6 figures, 2 tables, and 16 references of which 6 are Soviet, 8 English, and 2 German.

Zimin, A.V., Churmanteyev, S.V. The Effect of  $\gamma$ -Radiation on Benzene-Ammonia Mixture

273

This paper considers the effect of dosage, temperature, and aggregate state on the yield of aniline, hydrogen, and nitrogen. The activity of the  $\text{Co}^{60}$  radiation source was  $\sim 80$  and  $\sim 700$  g-equiv., and the period of irradiation was 3 to 256 hours. The basic products of radiolysis in the absence of air were aniline, hydrogen, nitrogen, and

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polymerization products. Dosage and temperature increases (to +90°) have no significant effect on the yield of aniline. The benzene - ammonia mixture irradiated in the presence of oxygen shows a sixfold increase in aniline yield. The greater amount of radicals is due to decrease in the recombination of primary products of radiolysis, since atomic hydrogen is bound by molecular oxygen. There are 4 tables, 2 figures, and 8 references of which 3 are Soviet, 3 English, and 2 French.

Zimin, A.V., Verina, A.D. The Effect of  $\gamma$ -Radiation on the Mixture of Benzene With Carbon Tetrachloride

280

The mixture was irradiated with  $Co^{60}$   $\gamma$ -radiation (activity 80, 130, and 1450 g-equiv.). The authors studied the effect of the dosage and molar ratio of components on the yield of HCl and of the non-volatile residue. Variation of the dosage had no effect and the ratio variation of  $C_6H_6 : CCl_4$  from 4 : 1 to 1 : 4 only slightly changed the results. The nonvolatile residue contains 70 percent of condensation products,  $\sim$  20 percent of high boiling fractions, and  $\sim$  10 percent of benzene. The most probable products of the high boiling fractions are monochlorobenzotrachloride isomers with a yield of  $\sim$  0.7 molecules per 100 ev. There are 3 tables and

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3 references of which 1 is Soviet and 2 German.

Vereshchinskiy, I.V., Karpushkin, L.T. Effect of  $\gamma$ -Radiation on the Synthesis of Indophenol

285

The synthesis of indophenol from an irradiated mixture of  $\alpha$ -naphthol and *n*-phenylene diamine was studied. The radiochemical yield of indophenol is 4.7 molecules per 100 ev, and the yield of the leuco form from the dye solution in vacuum is  $\sim$  1.8 molecules per 100 ev. It was shown that the formation of the dye is a two-stage process. The first stage is the formation of the leuco form. It does not require the presence of molecular oxygen. The second stage is the conversion of the leuco form into the dye. This stage occurs only in the presence of molecular oxygen. Irradiation considerably accelerates the conversion of the leuco form. There are 9 figures and 12 references of which 4 are Soviet, 6 German, 1 English, and 1 French.

PART 4. EFFECT OF RADIATION ON POLYMERS

Slovokhotova, N.A. Infrared Spectroscopic Study of Chemical Changes Occurring in Polytetrafluoroethylene (Teflon) Due to Ionizing Radiation

295

The infrared spectra of Teflon indicated that the conjugate

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Effect of Ionizing Radiation (Cont.)

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C — C double bonds form during irradiation with fast electrons and with  $\text{Co}^{60}$   $\gamma$ -radiation. Irradiation in air leads to the interaction of teflon with water vapor with the formation of C — O, OH, and CH groups. Irradiation converts crystalline teflon into amorphous teflon and possibly causes the formation of perfluorocyclobutene rings. There are 7 figures and 30 references, of which 3 are Soviet, 26 English, and 1 French.

Taubman, A.B., Yanova, L.P. Study of the Radiation Stability of High Polymers. Part 1. Effect of Radiation on Diffusion Permeability

307

The effect of radiation on diffusive permeability of high-polymer films is conditioned by the physical state and structure of the polymers. Vitreous polyvinylmethacrylate with low permeability to gases shows a sharp increase in permeability after relatively small radiation doses. This is explained by the radiational instability of the densely packed macromolecules. The coefficients of permeability and diffusion through polyvinyl chloride and polyethylene were computed for hydrogen chloride. There are 3 figures, 2 tables and 9 references of which 3 are Soviet and 6 English.

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Effect of Ionizing Radiation (Cont.)

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Yanova, L.P., Taubman, A.B. Study of the Radiation Stability of High Polymers. Part 2. Role of Gas Formation in the Destruction of Polymers 314

The role of gas formation was studied in the destruction of polytetrafluorethylene, polymethylmethacrylate and polyethylene due to electron irradiation. The intensity of the process of formation of gaseous destruction products increases sharply in a narrow temperature range of polymer fusion or transition to visco-fluid state. It was shown that the destruction should be regarded as a reversible process: destruction  $\leftrightarrow$  recombination of free radicals. The equilibrium shifts to the left with transition of the polymer into the fluid state. Changes in the mechanical properties of polymers and finally their destruction, are determined by the ratio of decomposition rates to gas formation rates, and, therefore, depends on the permeability to gases of the sample subjected to fast-electron irradiation. Crack formation in the polymer due to irradiation is traced to gas formation and emanation. There are 8 figures, 3 tables, and 9 references of which 6 are Soviet, and 3 English.

Kargin, V.A., Taubman, A.B., Yanova, L.P., Belyayeva, Z.F. Effect of Ionizing Radiation on the Properties of Vinyl Chloride and Vinylidene Chloride Copolymers 325

The effect of irradiation on gas permeability and mechanical

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Effect of Ionizing Radiation (Cont.)

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properties of the copolymers is connected with changes in the mixed amorphous - crystalline states and microstructure of the copolymers. The presence of crystallizing components increase the microdefects which results in increased permeability to gases. A reverse process can be observed for a specific range of small doses when the crystalline component undergoes fusion and the material becomes amorphous. Therefore, the curves for the permeability coefficient as a function of dosage have a maximum and a minimum. There are 8 figures, 2 tables, and 7 references of which 6 are Soviet, and 1 French.

Nikitina, T.S., Kuz'minskiy, A.S., Karpov, V.L. Radiation Vulcanization of Rubber

335

This is a study of radiation vulcanization of pure rubbers: natural rubber (NK), butadiene-styrene rubber (SKB-30) butadiene rubber (SKB) and butadiene-nitril rubber (SKN-26). The tensile strength of the radiation vulcanizates of NK does not exceed 30kg/cm. Sulfur and certain sulfur compounds (tetramethylthiuram-disulfide) lower the rate of the vulcanization. Compounds containing heavy atoms (zinc oxide, kaolin, chalk) accelerate the vulcanization only when X-rays are used. Gas and lamp blacks are very active fillers.

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790

The effect of carbon blacks is modified by their degree of oxidation. There are 8 figures, 1 table, and 17 references of which 7 are Soviet, 8 English, 1 French, and 1 German.

Tsetlin, B.L., Sibirskaya, G.K. Effect of Ionizing Radiation on the Thermomechanical Properties of Polyethylene

344

Polyethylene undergoes vulcanization when irradiated with electrons and X-rays. At a temperature corresponding to the melting point of the initial polymer, the irradiated substance converts to a highly elastic state the modulus of which is proportional to the dosage. In the range of  $10^{16}$  -  $10^{19}$  ev/cm sec vulcanization effectiveness follows the principle of equivalent effect of equal doses. The thermal stability of polyethylene increases with increased dosage. The mechanism of radiation vulcanization is related to the recombination of the  $R_1\dot{C}H \cdot R_2$  radicals which result from primary rupture of the C - H bonds. There are 4 figures, 1 table, and 22 references of which 14 are Soviet, 7 English, and 1 French.

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